

# Master JEE CLASSES Kukatpally, Hyderabad.

# **IIT-JEE-MAINS PAPER-6**

Max.Marks:360

#### **IMPORTANT INSTRUCTIONS:**

- 1) Immediately fill in the particulars on this page of the Test Booklet with Blue/Black Ball Point Pen. Use of pencil is strictly prohibited.
- 2) The test is of 3 hours duration.
- The Test Booklet consists of 90 questions. The maximum marks are 360.
- 4) There are three parts in the question paper A, B, C consisting of Mathematics, Physics and Chemistry having 30 questions in each part of equal weight age. Each question is allotted 4 (four) marks for correct response.
- Candidates will be awarded marks as stated above in instruction No. 4 for correct response of each question. (1/4) (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from the total score will be made if no response is indicated for an item in the answer sheet.
- 6) There is only one correct response for each question. Filling up more than one response in any question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instruction 5 above.

#### **SYLLABUS**

#### MATHS:

Trigonometric Equations (30%); Determinant (70%) **PHYSICS:** 

Projectile motion on inclined planes, relative velocity (Relative velocity 1D,2D,Rate of approach and rate of separation, condition for collision of two projectiles, relative motion of one projectile with respect to another projectile,rain problems,river boat problems,air craft wind problems etc..) **CHEMISTRY:** 

Chemical Equilibrium: Equilibrium involving physical processes: solid - liquid; liquid - gas and solid - gas. General characteristics of equilibrium involving physical process, Equilibrium involving chemical processes, characteristics of equilibrium state, Law of mass action, equilibrium constant, characteristics of equilibrium constant, Kp, Kcrelation, Lechatlier's principle - factors effecting equilibrium concentration, pressure, temperature and effect of catalyst, principle applied for synthesis of ammonia and sulphur trioxide and other typical reactions, Thermodynamics of equilibrium constant. Significance of  $\Delta G$  and  $\Delta G^{\circ}$  in chemical equilibrium(70%) Enthalpy of reaction - different types of enthalpies, Hess's law, Effect of temperature on enthalpy of reaction - Kirchoff's equation, Hess's law and its applications, Determination of enthalpy of hydration of CuSO4, Resonance energy,Limitations of first law of thermodynamics; Second law of Thermodynamics; Carnot cycle, efficiency of thermodynamic system, Entropy - physical significance, expressions in all types of processes ,Gibb's free energy - physical significance, spontneity of processes(30%)

MATHS		$\begin{vmatrix} x & x+y & x+y+z \end{vmatrix}$
1. For all values of $\theta$ ,	4.	If $\begin{vmatrix} 2x & 3x + 2y & 4x + 3y + 2z \end{vmatrix} = 64$ ,
		$\begin{vmatrix} 3x & 6x + 3y & 10x + 6y + 3z \end{vmatrix}$
$\sin\theta$ $\cos\theta$ $\sin2\theta$		then v-
$\sin\left(\theta + \frac{2\pi}{3}\right) \cos\left(\theta + \frac{2\pi}{3}\right) \sin\left(2\theta + \frac{4\pi}{3}\right) =$		
$\sin\left(\theta - \frac{2\pi}{2}\right) \cos\left(\theta - \frac{2\pi}{2}\right) \sin\left(2\theta - \frac{4\pi}{2}\right)$		1) 1 2) 4
$\left  \operatorname{Sin} \begin{pmatrix} 0 & 3 \\ 3 \end{pmatrix} \right  = \left  \operatorname{Sin} \begin{pmatrix} 2 & 3 \\ 3 \end{pmatrix} \right $		3) -1 4) 2
1) 0 2) 1		-, - , -
3) 2 4) -1	5.	The number of positive integral solutions
2. If $a \neq p$ , $b \neq q$ , $c \neq r$ and $\begin{vmatrix} p & b & c \\ a & q & c \\ a & b & r \end{vmatrix} = 0$ .		of the equation $\begin{vmatrix} x^3 + 1 & x^2y & x^2z \\ xy^2 & y^3 + 1 & y^2z \\ xz^2 & yz^2 & z^3 + 1 \end{vmatrix} = 11$
Then find the value of		ia
p + q + r =		15
$\overline{p-a}$ $\overline{q-b}$ $\overline{r-c}$		1) 0 2) 6
1) 1 2) 2		3) 3 4) 12
3) 3 4) 4	6.	The value of
3. The maximum value of $\begin{vmatrix} 1+\sin^2 x & \cos^2 x & 4\sin 2x \\ \sin^2 x & 1+\cos^2 x & 4\sin 2x \\ \sin^2 x & \cos^2 x & 1+4\sin 2x \end{vmatrix}$ is		$\begin{vmatrix} a^2 & a & 1\\ \cos nx & \cos(n+1)x & \cos(n+2)x\\ \sin nx & \sin(n+1)x & \sin(n+2)x \end{vmatrix}$
· · ·		independent of
1) 0 2) 2		1) n 2) a
3) 4 4) 6		3) x 4) None of these
space for r	l ough wo	ork Page 2

7.	The value of the determinant	10.	If the determinant
	$\begin{vmatrix} a-b & b-c & c-a \\ x-y & y-z & z-x \\ p-q & q-r & r-p \end{vmatrix}$ is: 1) 0 2) $abc + pqr + xyz$		$\begin{vmatrix} b-c & c-a & a-b \\ b'-c' & c'-a' & a'-b' \\ b''-c'' & c''-a'' & a''-b'' \end{vmatrix} = m \begin{vmatrix} a & b & c \\ a' & b' & c'' \\ a'' & b'' & c'' \end{vmatrix}$ where $\begin{vmatrix} a & b & c \\ a' & b' & c'' \\ a'' & b'' & c'' \end{vmatrix} \neq 0$
	3) $(a-x)(y-z)(r-p)$		then the value of 'm' is
	4) <i>abc</i>		1) 0 2) 2
8.	If $\begin{vmatrix} x & 3 & 6 \\ 3 & 6 & x \\ 6 & x & 2 \end{vmatrix} = \begin{vmatrix} 2 & x & 7 \\ x & 7 & 2 \\ 7 & 2 & x \end{vmatrix} = \begin{vmatrix} 4 & 5 & x \\ 5 & x & 3 \\ x & 4 & 5 \end{vmatrix} = 0$		3) -1 4) 1
	$\begin{vmatrix} 6 & x & 5 \end{vmatrix} \begin{vmatrix} 7 & 2 & x \end{vmatrix} \begin{vmatrix} x & 4 & 5 \end{vmatrix}$ then x is equal to	11.	If $D_1 = \begin{vmatrix} 1 & 1 & 1 \\ x^2 & y^2 & z^2 \\ x & y & z \end{vmatrix}$ and $D_2 = \begin{vmatrix} 1 & 1 & 1 \\ yz & xz & xy \\ x & y & z \end{vmatrix}$ ,
	1) 9 2) -9		then
	3) 0 4) 5		1) $D_1 = D_2$ 2) $D_1 = -D_2$
9.	If		3) $D_1 = -2D_2$ 4) $D_2 = 2D_1$
$\begin{vmatrix} x^3 + x^3 \\ x - x^3 \end{vmatrix}$	$\begin{vmatrix} 4x & x+3 & x-2 \\ -2 & 5x & x-1 \\ -3 & x+2 & 4x \end{vmatrix} = ax^5 + bx^4 + cx^3 + dx^2 + ex + f,$ be an identity in x, where a,b,c,d,e, f	$\begin{vmatrix} 12. \\ a + \end{vmatrix}$	If b+2c $a$ $bc b+c+2a b =K(a+b+c)^3,$
	are independent of $x$ , then the value		c $a$ $c+a+2b$
	of f is		then K is equal to
	1) 0 2) 15		1) 1     2) 2
	3) 17 4) 13		3) 4 4) 6
	space for roug	h wo	ork Page 3

13. If [] denote the greatest integer less  
than or equal to the real number  
consideration and  
$$-1 \le x < 0, 0 \le y < 1, 1 \le z < 2$$
, then the  
value of the determinant  
 $\begin{bmatrix} x \\ 1 + 1 \end{bmatrix} \begin{bmatrix} y \\ z \end{bmatrix} \begin{bmatrix} z \\ x \end{bmatrix} \begin{bmatrix} y \\ y \end{bmatrix} + 1 \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ x \end{bmatrix} \begin{bmatrix} y \\ y \end{bmatrix} + 1 \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ x \end{bmatrix} \begin{bmatrix} y \\ z \end{bmatrix} + 1 \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} y \\ z \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix} \begin{bmatrix} z \\ z \end{bmatrix} \end{bmatrix}$ 

$$x_1$$
 $y_1$ 119. $\begin{vmatrix} x_1 + bx_2 + cx_3 & ay_1 + by_2 + cy_3 & a + b + c \\ -ax_1 + bx_2 + cx_3 & -ay_1 + by_2 + cy_3 & -a + b + c \end{vmatrix}$ 22. The general solution of the equation1) 0 $\sin^{50} x - \cos^{50} x = 1 is$ 1)  $2n\pi + \frac{\pi}{2}$ 2)  $2n\pi + \frac{\pi}{3}$ 2) 1 $3) a(x_1 + x_2 + x_3)$ 3)  $n\pi \pm \frac{\pi}{2}$ 4)  $n\pi \pm \frac{\pi}{3}$ 3)  $a(x_1 + x_2 + x_3)$  $b(y_1 + y_2 + y_3) + c$ 23. General solution of the equation20. The number of values of  $\theta$  in  $\left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$ such that  $\theta \neq \frac{n\pi}{5}$  for  $n = 0, \pm 1, \pm 2$  and23. General solution of the equation $(\sqrt{3} - 1)\sin \theta + (\sqrt{3} + 1)\cos \theta = 2$  is1)  $2n\pi \pm \frac{\pi}{4} + \frac{\pi}{12}$ 2)  $n\pi + (-1)^n \cdot \frac{\pi}{4} + \frac{\pi}{12}$ 21. The equation $2 \ln \pi \pm \frac{\pi}{4} + \frac{\pi}{12}$ 2)  $n\pi + (-1)^n \cdot \frac{\pi}{4} - \frac{\pi}{12}$ 23. The equation $2 \cos^2 \frac{x}{2} \cdot \sin^2 x = x^2 + \frac{1}{x^2}; 0 \le x \le \frac{\pi}{2}$  has.1)  $0$  ne real solution24. The solution of  $\sin^8 x + \cos^8 x = \frac{17}{32}$  is25. Total number of solution26. No solution27. No solution28. No solution29. No solution3) More than one real solution4) None of these1)  $2 - x \ge 4$ 2)  $2 - x + \frac{\pi}{3}$ 3)  $6 - 4 > 8$ 

26. For 
$$0 < \theta < \pi/2$$
, the number of  
solutions of29. If the equation $\sum_{m=1}^{6} \cos ec \left(\theta + \frac{(m-1)\pi}{4}\right) \cos ec \left(\theta + \frac{m\pi/4}{4}\right) = 4\sqrt{2}$   
is10.  $(\cos p) - 1)x^2 + (\cos p)x + \sin p = 0$  in the  
variable x has real roots then p can take  
any value in the interval  
1) 1 2) 0  
3) 2 4) 310.  $(\cos p) - 1)x^2 + (\cos p)x + \sin p = 0$  in the  
variable x has real roots then p can take  
any value in the interval  
1)  $(0, 2\pi)$  2)  $(-\pi, 0)$ 27. The most general values of  $\theta$  which  
satisfies both the equations  $\tan \theta = -1$   
and  $\cos \theta = \frac{1}{\sqrt{2}}$  is  
1)  $n\pi + 7\frac{\pi}{4}$  2)  $2n\pi + \frac{7\pi}{4}$   
3)  $n\pi + (-1)^{\nu} \frac{7\pi}{4}$  4)  $\frac{7n\pi}{4}$   
28. If  $\cos \theta \cos 2\theta \cos 3\theta = \frac{1}{4}$  for  $0 < \theta < \pi$   
then  $\theta =$   
1)  $\frac{\pi}{7}, \frac{5\pi}{7}, \pi$   
2)  $\frac{\pi}{2}, \frac{\pi}{4}, \frac{3\pi}{4}, \frac{\pi}{6}, \frac{5\pi}{6}$   
3)  $\frac{\pi}{8}, \frac{3\pi}{8}, \frac{5\pi}{8}, \frac{7\pi}{8}, \frac{2\pi}{3}, \frac{\pi}{3}$   
4)  $\frac{2n\pi}{3}$  :  $n \in \mathbb{Z}$  or  $n\pi + \frac{\pi}{4}$  :  $n \in \mathbb{Z}$ 29. If the equation  
(( $\cos p) - 1$ )x<sup>2</sup> + ( $\cos p$ )x +  $\sin p = 0$  in the  
variable x has real roots then p can take  
any value in the interval  
1)  $(0, 2\pi)$  2)  $(-\pi, 0)$   
3)  $\left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$   
4)  $(0, \pi)$ 30.  $\frac{\sin^3 \theta - \cos^3 \theta}{\sin \theta - \cos \theta} - \frac{\cos \theta}{\sqrt{1 + \cot^2 \theta}} - 2\tan \theta \cot \theta = -1$   
if  
1)  $\theta \in \left(0, \frac{\pi}{2}\right)$  2)  $\theta \in \left(\frac{\pi}{2}, \pi\right)$   
3)  $\theta \in \left(\pi, \frac{3\pi}{2}\right)$  4)  $\theta \in \left(\frac{3\pi}{2}, 2\pi\right)$ 21.  $\frac{\pi}{3}, \frac{3\pi}{8}, \frac{5\pi}{8}, \frac{7\pi}{8}, \frac{2\pi}{3}, \frac{\pi}{3}$   
4)  $\frac{2n\pi}{3}$  :  $n \in \mathbb{Z}$  or  $n\pi + \frac{\pi}{4}$  :  $n \in \mathbb{Z}$ space for rough work

#### PHYSICS

31. A ball is thrown at speed v from zero height on a level ground. At what angle with horizontal should the ball be thrown so that the area under the trajectory is maximum.

 1) 30°
 2) 45°

 3) 60°
 4) 37°

32. A ball falls from height h. It bounces off a surface at height y(with no loss in speed). The surface is inclined at 45°, so that the ball bounces off horizontally. What should y be so that the ball travels a maximum horizontal distance?

1) 
$$\frac{h}{2}$$
 2)  $\frac{h}{3}$   
3)  $\frac{2h}{3}$  4)  $\frac{h}{4}$ 

33. Point A moves uniformly with velocity v so that the vector  $\vec{v}$  is continually 'aimed' at point B which in its turn moves recti-linearly and uniformly with velocity u<v. At the initial moment of time  $\vec{v} \perp \vec{u}$  and the points

are separated by a distance *l*. How soon will the points converge?

1) 
$$\frac{vl}{\left(v^2 - u^2\right)}$$
  
2)  $\frac{ul}{\left(v^2 - u^2\right)}$   
3)  $\frac{ul}{\left(u^2 - v^2\right)}$   
4)  $\frac{ul}{\left(u^2 + v^2\right)}$ 

- 34. A ball is thrown from the edge of a cliff of height h=14.4 m with a speed of 12m/s. At what inclination angle with horizontal should it be thrown so that it travels a maximum horizontal distance when it hits the ground? Assume that the ground is below the cliff level.(g=10m/s<sup>2</sup>) 1) 45° 2) 60°
  - 3) 30° 4) None of these
- 35. A, B & C are three objects each moving with constant velocity. A's speed is 10 m/sec in a direction PQ. The velocity of B relative to A is 6 m/sec at an angle of, cos<sup>-1</sup> (15/24) to PQ. The velocity of C relative to B is 12 m/sec in a direction QP, then find the magnitude of the velocity of C
  1) 6 m/sec
  2) 4 m/sec
  3) 5 m/sec
  4) 10 m/sec

space for rough work

36. Two particles A and B start moving from the same point on the X-axis. The velocity versus time graph for the particle is as shown in figure. The maximum relative separation between the two particles will be equal to



37. Three particles are initially kept at the vertices A, B and C of a triangle with  $|\underline{A} = 120^{\circ}$  and  $|\underline{B} = |\underline{C} = 30^{\circ}$ . They simultaneously start moving and will simultaneously meet at the

incentre(point of intersection of angular bisectors) of  $\Delta ABC$  and then stop. The ratio of average velocities of the particles is

- 1)  $sin15^{\circ}: cos 30^{\circ}: cos 30^{\circ}$
- 2)  $\cos 15^\circ : \sin 30^\circ : \sin 30^\circ$
- 3) tan15°: tan 30°: tan 30°
- 4)  $\cot 15^\circ : \cot 30^\circ : \cot 30^\circ$
- 38. Two particles start simultaneously from the same point and move along two straight lines, one with uniform velocity v and other with a uniform acceleration a. If  $\alpha (< 90^{\circ})$  is the angle between the lines of motion of two particles then the least value of relative velocity will be at time
  - $1)\left(\frac{v}{a}\right)\sin\alpha \qquad 2)\left(\frac{v}{a}\right)\cos\alpha$

3)
$$\left(\frac{v}{a}\right)$$
tan  $\alpha$  4) $\left(\frac{v}{a}\right)$ cot  $\alpha$ 

39. Two inclined planes (I) and (II) have inclination  $\alpha$  and  $\beta$  respectively with horizontal, (where  $\alpha + \beta = 90^{\circ}$ ) intersect each other at point O as shown in figure.



A particle is projected from point A with velocity u along a direction perpendicular to plane (I). If the particle strikes plane (II) perpendicular at B, then



- 1) time of flight =  $\frac{u}{g \sin \beta}$
- 2) time of flight =  $\frac{u}{g \sin \alpha}$
- 3) distance  $OB = \frac{u^2}{g \sin \beta}$

4) distance 
$$OB = \frac{u^2}{2g\sin\alpha}$$

40. A 2m wide truck is moving with a uniform speed  $v_0 = 8$  m/s along a straight horizontal road. A pedestrian starts to cross the road with a uniform speed v when the truck is 4 m away from him. The minimum value of v so that he can cross the road safely is



41. The front windscreen of a car is inclined at an angle 30° with the horizontal. Hailstones fall vertically downwards with a speed of  $5\sqrt{3}$  m/s. The speed of the car so that hailstones are bounced back by the screen in vertically upward direction with respect to car is (Assume elastic collision of hailstones with wind screen)

1) 20 m/s	2) 10 m/s
3) 15 m/s	4) 5 m/s

42. A particle is projected from point A at an angle of <sup>53°</sup> with horizontal. At the same time wedge starts from rest and moves with constant acceleration a as shown. The value of a for which the particle

space for rough work



44. A ball 'A' is thrown vertically up with a speed of 24 m/s and at the same time

another ball 'B' is thrown horizontally from a tower as shown. Assuming 2D motion find the speed of the ball B which will ensure that ball's will be collide in air.



45. The current velocity (V) of a river is directly proportional with the distance (x) from its bank V is maximum in the middle and equals to  $v_0 > .$  A boat is moving on the river with a constant velocity  $\mu$  relative to the water and perpendicular to the current. Find the drift of the boat. Width of the river is 'C'

$$1)\frac{v_0C}{2\mu} \qquad 2)\frac{v_0C}{4\mu} \\ 3)\frac{v_0C}{\mu} \qquad 4)\frac{v_0C}{3\mu}$$

space for rough work

46. A boat can travel at a speed of  $3ms^{-1}$ on still water. A boatman wants to cross the river while covering the shortest possible distance. In what direction should he row with respect to the bank, if the speed of the water is  $4ms^{-1}$ ?

1) making an angle of  $\tan^{-1}\left(\frac{4}{3}\right)$  with

the bank and he has to row down stream

2) making an angle of  $\sin^{-1}\left(\frac{3}{4}\right)$  with the bank and he has to row down stream

3) making an angle of  $\cos^{-1}\left(\frac{3}{4}\right)$  with

the bank and he has to row up stream 4) making an angle of  $\tan^{-1}\left(\frac{3}{4}\right)$  with

the bank and he has to row up stream

47. Twelve persons are initially at the twelve corners of a regular polygon of twelve sides of side a. Each person now moves with uniform speed V in such a manner that 1 is always directed towards 2, 2 towards 3, 3 towards 4 and so on. The distance travelled by each person before they meet is

1)
$$\frac{2a}{2+\sqrt{3}}$$
 2) $\frac{2a}{2-\sqrt{3}}$   
3) $\frac{2a}{\sqrt{3}}$  4) $\frac{a}{2+\sqrt{3}}$ 

- 48. A boat is moving towards east with velocity 4 m/s with respect to still water and river is flowing towards north with velocity 2 m/s and the wind is blowing towards north with velocity 6 m/s. The direction of the flag blown over by the wind hoisted on the boat is:
  - 1) north-west
  - 2) south-east
  - 3)  $\tan^{-1}(1/2)$  with east
  - 4) north
- 49. An airplane pilot wants to fly from city A to city B which is 1000 km due north of city A. The speed of the plane in still air is 500 km/hr. The pilot neglects the effect

space for rough work

of the wind and directs his plane due north and 2 hours later find himself 300km due north-east of city B. The wind velocity is 1) 150km/hr at  $45^{\circ}$ N of E 2) 106km/hr at 45°N of E 3) 150 km/hr at  $45^{\circ}$ N of W 4) 106 km/hr at 45°N of W 50. A boat which has a speed of 5 km/h in still water crosses a river of width 1 km along the shortest possible path (zero drift) in 15min. The velocity of the river water in km per hour is 2) 3 1)1 4)  $\sqrt{41}$ 3)4 51. In a lake, stream flow direction is shown in the figure which is same every where. A man starting from the

point P on the bank wants to move to the

other bank in shortest time. He should

swim



1) along PQ

2) along PR

time

3) in a direction in-between PQ and PR4) in all cases he would reach at the same

52. A boat is moving with velocity v = 2m/sw.r.t. river at an angle  $\theta = 30^{\circ}$ . Velocity of river is u = 4 m/s. A man at the bank is moving with velocity  $v_0$  as shown in figure. Boat is moving perpendicularly to the bank w.r.t. the man then  $v_0$  is

space for rough work



53. A number of projectiles each with same speed u are fired at different angles lying between 0° and 90° as shown. Neglecting air resistance and assuming g to be constant, then the equation of the envelop E of the parabolic tractories (as shown in figure) is



1) 
$$y = \frac{u^2}{2g} - \frac{gx^2}{2u^2}$$
  
2)  $y = \frac{gx^2}{2u^2} + \frac{u^2}{2g}$   
3)  $y = \frac{2u^2}{gx^2} - \frac{2g}{u^2}$   
4)  $y = \frac{2u^2}{gx^2} + \frac{2g}{u^2}$ 

- 54. An airplane has to fly along a straight path from town A to town B, 500 km away. Town B is due east of town A and a strong wind blows from north to south at 300 km/hr. If the plane's airspeed is 900 km/hr, which of the following statements is true?
  - 1) Trip time is  $\frac{5}{3\sqrt{8}}$  hr.
  - 2) Plane's speed with respect to ground is600 km/hr.
  - 3) Plane's heading is  $30^{\circ}$  North of East.
  - 4) None of the above
- 55. A boat is to go straight across a stream

that is flowing at 5.0 km/h to east. The

space for rough work



58. A man is coming down an incline of angle 30°. When he walks with speed  $2\sqrt{3}$  m/s he has to keep his umbrella vertical to protect himself from rain. The actual speed of rain is 5 m/s. At what angle with vertical should he keep his umbrella when he is at rest so that he does not get drenched?



59. If  $\vec{r}$  is the position vector of a particle

and  $\vec{v}\,$  its velocity then

- 1)  $\hat{v} \cdot \vec{r} = \frac{dr}{dt}$  2)  $\vec{v} \cdot \hat{r} = \frac{dr}{dt}$
- $3)\vec{v}\cdot\vec{r}=0 \qquad \qquad 4) \vec{v}\cdot\vec{r}=vr$

60. A man is standing on incline which makes θ with horizontal and rain is falling vertically with respect to man. Now man starts walking along the incline. Mark the correct option.



 If man walks up the incline, rain may appear to come horizontally
 If man walks down the incline, rain may appear to come horizontally
 No matter whether man walks up or down, rain can never appear to come horizontally.

4) If man walks (up or down) along

incline rain may appear to come along the incline.

space for rough work

	CHEMISTRY		The equilibrium constant of the reaction
61.	In the reaction $CS_2(l) + 3O_2(g) \rightarrow$		$NH_3 + \frac{1}{2}O_2 \rightleftharpoons NO + \frac{3}{2}H_2$ in terms of K <sub>1</sub> ,
	$CO_2(g) + 2SO_2(g) \Delta H = -275 \text{ kcal}$		$2$ $2$ $K_2$ is:
	The enthalpies of formation of SO <sub>2</sub> and		$1$ $\sqrt{K_2}$
	$CS_2$ are -75 and +25kcal/mol		1) $\frac{1}{K_1 K_2}$ 2) $\frac{\sqrt{2}}{K_1}$
	respectively. Calculate the enthalpy of		3) $1$ (1) $K_2$
	formation of CO <sub>2</sub> .		$\sqrt{K_1 K_2}$ $\sqrt{K_1}$
	1) – 75 kcal/mol 2) – 50 kcal/mol	64.	Certain amount of PCl <sub>5</sub> is taken in a
	3) – 100 kcal/mol 4) – 150 kcal/mol		closed container to establish the
62.	For the reaction $Br_2(g) \rightarrow 2Br(g)$ ,		equilibrium $PCl_5 \rightleftharpoons PCl_3 + Cl_2, 25\%$ of
	$\Delta$ H = 'a' KJ the correct statement (s)		PCl <sub>5</sub> is found to be dissociated at
	is/are		equillibrium. If total pressure at
	1) Enthalpy of formation of Br(g) is		equilibrium is 30 atm then find $k_p$ for the
	'a' KJ		reaction
	2) Enthalpy of atomisation of Br(g) is		1) 4 2) 1/2
	'a' KJ		3) 2 4) 1/4
	3) Bond enthalpy of Br–Br bond is	65.	For the reaction $CaCO_3(S) \rightleftharpoons CaO(s) +$
	'a' KJ		$CO_2(g)$ , the pressure of $CO_2(g)$ depends
	4) Bond enthalpy of Br-Br bond is 'a'		on
	KJ but enthalpy of atomisation of		1) The mass of $CaCO_3(s)$
	Br <sub>2</sub> (g) is not equal to 'a' KJ		2) The mass of CaO(s)
63.	The following equilibria are given :		3) The masses of both $CaCO_3(s)$ and
	$N_2 + 3H_2 \rightleftharpoons 2NH_3 - K_1$		CaO(s)
	$N_2 + O_2 \rightleftharpoons 2NO \longrightarrow K_2$		4) Temperature of the system

space for rough work

66.	Which of the reaction defines $\Delta \mathbf{H}^{\mathbf{o}}_{f}$		+2 <i>H</i>
	1) $C_{(diamond)} + O_2(g) \rightarrow CO_2(g)$		ii) <i>N</i>
	2) $\frac{1}{2}H_2(g) + \frac{1}{2}F_2(g) \rightarrow HF(g)$		iii) 2
	3) $N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$		1) ii
	1		3) i a
	4) $CO(g) + \frac{1}{2}O_2(g) \rightarrow CO_2(g)$	69.	For
67.	In Haber's process of ammonia		$X_2C$
	manufacture :		$\Delta E =$
	$N_2(g) + 3H_2(g) \rightarrow 2NH_3(g);$		Hen
	$\Delta H_{25}^{0}{}_{0} = -92.2kJ$		1) 2
	$Melagula = N(\alpha) + L(\alpha) + NH(\alpha)$		3) 9.
	$C \ IK^{-1}mol^{-1} \ 20 \ 1 \ 28 \ 8 \ 25 \ 1$	70.	If $\Delta$
	CpJK 1101 29.1 20.0 55.1		(1) /
	If $C_P$ is independent of temperature,		(1)
	then reaction at 100°C as compared to		and
	that of 25°C will be :		(2)
	1) More endothermic		X 71
	2) Less endothermic		Wha
	3) More exothermic		form
	4) Less exothermic		1) +
68.	Predict which of the following reaction		
	(s) has a negative entropy change?		3) -
	i) $CH_4(g) + 2O_2(g) \rightarrow CO_2(g)$		

 $H_2O(l)$  $NH_3(g) + HCl(g) \rightarrow NH_4Cl(s)$  $2KClO_4(s) \rightarrow 2KClO_3(s) + O_2(s)$ i 2) ii and ii 4) i the reaction  $D_4(l) \longrightarrow 2XO_2(g)$  $= 2.1 \ kcal. \ \Delta S = 20 \ cal \ / \ K$  at 300 K ce  $\Delta G$  is 2) -2.7 kcal .7 kcal .3 kcal 4) -9.3 kcal G = -177 K cal for  $2Fe(s) + \frac{3}{2}O_2(g) \rightarrow Fe_2O_3(s)$  $\Delta$  G=19 K cal for  $4Fe_2O_3(s) + Fe(s) \rightarrow 3Fe_3O_4(s)$ at is the Gibbs free energy of nation of  $Fe_3O_4$ ?  $-229.6\frac{kcal}{mol} \qquad 2) -242.3\frac{kcal}{mol}$  $-727\frac{kcal}{mol} \qquad 4) -229.6\frac{kcal}{mol}$ 

space for rough work

- 71. When a certain amount of ethylene was combusted, 6226 kj heat was evolved. If heat of combustion of ethylene is 1411 kJ, the volume of O<sub>2</sub> (at NTP) that entered into the reaction is -
  - 1) 296.5ml
  - 2) 296.5 liters
  - 3)  $6226 \times 22.4$  liters
  - 4) 22.4 liters
- 72. 1 mole of an ideal gas at 25°C is subjected to expand reversibly ten times of its initial volume. The change in entropy due to expansion is :
  - 1) 19.15 JK<sup>-1</sup> mole<sup>-1</sup>
  - 2) 16.15 JK<sup>-1</sup> mole<sup>-1</sup>
  - 3) 22.15  $JK^{-1}$  mole<sup>-1</sup>
  - 4) None
- 73. Following reaction occurs at 25°C : 2NO (g, 1 × 10<sup>-5</sup> atm) + Cl<sub>2</sub> (g, 1 × 10<sup>-2</sup> atm)  $\rightleftharpoons$  2NOCl (g, 1 × 10<sup>-2</sup> atm)  $\triangle$  G<sup>O</sup> is-1) -45.65 KJ 2) -28.53 KJ
  - 3) -22.82 KJ 4) -57.06 KJ

74. For the reaction  $N_2O + O_2 \rightleftharpoons 2NO$ equilibrium constant  $K_c = 2$ . Degrees of dissociation of  $N_2$  and  $O_2$  are

1) 
$$\frac{1}{1+\sqrt{2}}, \frac{1}{1-\sqrt{2}}$$
  
2)  $\frac{1}{1-\sqrt{2}}, \frac{1}{1+\sqrt{2}}$   
3) Both are  $\frac{1}{1+\sqrt{2}}$   
4)  $\frac{2}{1+\sqrt{2}}, \frac{2}{1-\sqrt{2}}$ 

75. consider the reaction equilibrium,

$$2SO_{2(g)} + O_{2(g)} \rightleftharpoons 2SO_{3(g)}, \Delta H^0 = -198kJ.$$

On the basis of Le chatelier's principle, the condition favourable for the forward reaction is

- 1) Lowering of temperature as well as pressure
- 2) Increasing temperature as well as pressure
- 3) Lowering the temperature and

increasing the pressure

4) Any value of temperature and pressure

space for rough work

76.	For an equilibrium
	$H_2O(s) \rightleftharpoons H_2O(l)$ which of the
	following statements is true.
	1) The pressure changes do not affect
	the equilibrium
	2) More of ice melts if pressure on the
	system is increased
	3) More of liquid freezes if pressure on
	the system is increased
	4) The pressure changes may increase
	or decrease the degree of advancement
	of the reaction depending upon the
	temperature of the system
77.	A 10 litre box contains $O_3$ and $O_2$ at
	equilibrium at 2000 K. $K_p = 4 \times 10^{14}$ atm
	for $2O_3(g) \rightleftharpoons 3O_2(g)$ Assume that
	$P_{O_2} >> P_{O_3}$ and if total pressure is 8
	atm, then partial pressure of $O_3$ will be
	1) 8×10 <sup>-5</sup> atm 2) 11.3×10 <sup>-7</sup> atm
	3) $9.71 \times 10^{-6}$ atm 4) $9.71 \times 10^{-2}$ atm
78.	The degree of dissociation of SO <sub>3</sub> is $\alpha$
	at equilibrium pressure $P_0$

$K_p$ for $2SO_3(g) \rightleftharpoons 2SO_2(g) + O_2(g)$ is
1) $\left[ \left( P_0 \alpha^3 \right) / 2 \left( 1 - \alpha \right)^3 \right]$
$2)\left[\left(P_{0}\alpha^{3}\right)/\left(2+\alpha\right)\left(1-\alpha\right)^{2}\right]$
$3)\left[\left(P_{0}\alpha^{2}\right)/2\left(1-\alpha\right)^{2}\right]$

- 4) None of these
- 79. When 3 moles of ethyl alcohol are mixed with 3 moles of acetic acid, 2 moles of ester are formed at equilibrium according to the equation

$$CH_{3}COOH(l) + C_{2}H_{5}OH(l)$$
  
$$\rightleftharpoons CH_{3}COOC_{2}H_{5}(l) + H_{2}O(l) \text{ The}$$

value of the equilibrium constant for the reaction is

1) 4	2) 2/9
3) 2	4) 4/9

80.  $X_2(g) + Y_2(g) \rightleftharpoons 2XY(g)$  reaction was studied at a certain temperature. In the beginning 1 mole of  $X_2$  was taken in a one liter flask and 2 moles of  $Y_2$  was taken in another 2 liter flask and both these containers are connected so equilibrium can be established. What is

space for rough work

 $\left[N_2 O_4\right] = \left[NO_2\right] = 0.1M$ the equilibrium concentration of  $X_{2}$ III :  $K_c = Q$  when point D or F is reached and  $Y_2$ ? Given Equilibrium 1) I. II 2) II, III concentration of [XY] = 0.63) I, III 4) I, II, III 82. The reaction moles/liter.  $A(g) + B(g) \rightleftharpoons C(g) + D(g)$  is studied  $1)\left(\frac{1}{3}-0.3\right),\left(\frac{2}{3}-0.3\right)$ in a one litre vessel at  $250^{\circ}$ C. The initial 2)  $\left(\frac{1}{3} - 0.6\right)$ ,  $\left(\frac{2}{3} - 0.6\right)$ concentration of A was 3n and that of B was n. when equilibrium was attained, 3) (1-0.3), (2-0.3)equilibrium concentration of C was found 4) (1-0.6), (2-0.6)to the equal to the equilibrium concentration of B. What is the 81.  $N_2O_4(g) \rightleftharpoons 2NO_2(g), K_c = 4$ . This concentration of D at equilibrium? reversible reaction is studied 1) n/22) (3n - 1/2)graphically as shown in figure. Select 3) (n - n/3)4) n the correct statements out of I, II and 83. For the equilibrium  $CuSO_4.5H_2O(s) \rightleftharpoons$ Ш  $CuSO_4.3H_2O(s) + 2H_2O(g)$  $K_p = 2.25 \times 10^{-4}$  atm<sup>2</sup> and vapour pressure Ε Concentration of water is 22.8 Torr at 298 K.  $CuSO_{4}5H_{2}O(s)$  is efflorescent (i.e. loses water) when relative humidity is 1) Less than 33.3% I: Reaction quotient has maximum 2) Less than 50% value at point A 3) Less than 66.6% II : Reaction proceeds left to right at a 4) above 66.6% point when space for rough work Page 20

- 84. An equilibrium mixture in a vessel of capacity 100 litre contain 1 mol  $N_2$ , 2 mol  $O_2$  and 3 mol NO. number of moles of  $O_2$  to be added so that at new equilibrium the conc. Of NO is found to be 0.04 mol/lit.
  - 1) (101/18) 2) (101/9)
  - 3) (202/9) 4) None of these
- 85. Equilibrium constants are given (in atm) for the following reactions  $0^{0}C$  $SrCl_{2}.6H_{2}O(s) \rightleftharpoons SrCl_{2}.2H_{2}O(s) +$  $4H_{2}O(g)$   $K_{p} = 5 \times 10^{-12}$  $Na_{2}HPO_{4}.12H_{2}O(s) \rightleftharpoons$  $Na_{2}HPO_{4}7H_{2}O(s) + 5H_{2}O(g)$  $K_{p} = 2.43 \times 10^{-13}$  $Na_{2}SO_{4}.10H_{2}O(s) \rightleftharpoons Na_{2}SO_{4}(s) +$  $10H_{2}O(g)$   $K_{p} = 1.024 \times 10^{-27}$ The vaopur pressure of water at  $0^{0}C$  is 4.56 torr. Which is the most effective drying agent at  $0^{0}C$

4) all equally

3)  $Na_2SO_4$ 

- 86. Ammonia dissociates into N₂ and H₂ such that degree of dissociation α is very less than 1 and equilibrium pressure is P₀ then the value of α is [ if Kp for 2NH₃(g) ⇒ N₂(g) + 3H₂(g) is 27 × 10<sup>-8</sup> P₀²
  1) 10<sup>-4</sup>
  2) 4×10<sup>-4</sup>
  3) 0.02
  - 4) Can't be calculated
- 87. For a reaction:  $A \rightleftharpoons B$ , carried out at

27<sup>°</sup>C, the ratio of equilibrium concentrations of product to reactant changes by a factor of  $e^4$  for every 1) 1.2 kcal rise in  $\Delta G^{\circ}$ 

- 2) 1.2 kcal fall in  $\Delta G^0$
- 3) 2.4 kcal rise in  $\Delta G^{\circ}$
- 4) 2.4 kcal fall in  $\Delta G^{0}$

2) Na<sub>2</sub>HPO<sub>4</sub>.7H<sub>2</sub>O

1) SrCl<sub>2</sub>. 2H<sub>2</sub>O

space for rough work

88. When a mixture of  $N_2$  and  $H_2$  in the volume ratio of 1 : 5 is allowed to react at 700 K and  $10^3$  atm pressure, 0.4 mole fraction of NH<sub>3</sub> is formed at equilibrium. The  $K_p$  for the reaction  $N_2(g) + 3H_2(g) \rightleftharpoons 2NH_3(g)$ 1)  $2.6 \times 10^{-5} atm^{-2}$  2)  $2.6 \times 10^{-4} atm^{-2}$ 3)  $2.6 \times 10^{3} atm^{-2}$  4)  $5.1 \times 10^{-3} atm^{-2}$ 89. In the system,  $LaCl_3(s) + H_2O(g) +$ heat  $\rightleftharpoons LaClO(s) + 2HCl(g)$ , equilibrium is established. More water vapour is added to disturb the equilibrium. If the pressure of water vapour at new equilibrium is double of that at initial equilibrium, the factor to which pressure of HCl is changed is 2)  $\sqrt{2}$  times 1) 2 times 3)  $\frac{1}{\sqrt{2}}$  times 4) 4 times 90.  $PCl_5$  (molecular mass = M) dissociates into PCl<sub>3</sub> and Cl<sub>2</sub> as: PCl<sub>5</sub>(g)  $\rightleftharpoons$  $PCl_3(g) + Cl_2(g)$ . If the total pressure of the system at equilibrium is P and

the density is 'd' at temperature, T K. The degree of dissociation of PCl<sub>5</sub> may be represented as

1) 
$$\frac{PM}{dRT}$$
 2)  $\frac{PM}{dRT}$  -1  
3)  $\frac{dRT}{PM}$  -1 4)  $\frac{dRT}{PM}$ 

space for rough work



# Master JEE CLASSES

# Kukatpally, Hyderabad.

#### IIT-JEE-MAINS PAPER-6 Max. Marks: 360

## **KEY SHEET**

# MATHS

1	1	2	2	3	4	4	2	5	2	6	1
7	1	8	2	9	3	10	1	11	1	12	2
13	3	14	3	15	1	16	3	17	4	18	2
19	1	20	3	21	2	22	3	23	1	24	1
25	3	26	3	27	2	28	3	29	4	30	2

## PHYSICS

31	3	32	1	33	1	34	3	35	3	36	3
37	1	38	2	39	1	40	3	41	3	42	3
43	4	44	2	45	1	46	3	47	2	48	1
49	1	50	2	51	1	52	3	53	1	54	1
55	4	56	3	57	1	58	4	59	2	60	2

# CHEMISTRY

61	3	62	3	63	4	64	3	65	4	66	2
67	3	68	3	69	2	70	2	71	2	72	1
73	1	74	3	75	3	76	2	77	2	78	2
79	1	80	1	81	2	82	1	83	2	84	1
85	1	86	3	87	4	88	1	89	2	90	2

$$\begin{array}{|c|c|c|c|c|} \hline \textbf{SOLUTIONS:} \\ \underline{\textbf{MATHS}} \\ \hline \textbf{MATHS} \\ \hline \textbf{I.} & \left| \begin{array}{c} \sin\left(\theta + \frac{2\pi}{3}\right) & \cos\left(\theta + \frac{2\pi}{3}\right) & \sin\left(2\theta + \frac{4\pi}{3}\right) \\ \sin\left(\theta - \frac{2\pi}{3}\right) & \cos\left(\theta - \frac{2\pi}{3}\right) & \sin\left(2\theta - \frac{4\pi}{3}\right) \\ \end{array} \right| = 0[\textbf{R}_1 \rightarrow \textbf{R}_1 + \textbf{R}_2 + \textbf{R}_3] \\ \hline \textbf{sin} \left(\theta - \frac{2\pi}{3}\right) & \cos\left(\theta - \frac{2\pi}{3}\right) & \sin\left(2\theta - \frac{4\pi}{3}\right) \\ \end{array} \right| = 0[\textbf{R}_1 \rightarrow \textbf{R}_1 + \textbf{R}_2 + \textbf{R}_3] \\ \hline \textbf{Solution} = 0 \\ \hline \textbf{a} & \textbf{b} & \textbf{r} \\ \hline \textbf{sin} \left(\theta - \frac{2\pi}{3}\right) & \cos\left(\theta - \frac{2\pi}{3}\right) & \sin\left(2\theta - \frac{4\pi}{3}\right) \\ \end{array} \\ \hline \textbf{Solution} = 0 \\ \hline \textbf{p}_{-a} + \frac{\textbf{q}}{\textbf{q} - b} + \frac{\textbf{r}}{\textbf{r} - c} = 2 \\ \hline \textbf{R}_2 - \textbf{R}_1 + \cos^2 x & 4\sin 2x \\ \hline \textbf{sin}^2 x & \cos^2 x & 4\sin 2x \\ \sin^2 x & 1 + \cos^2 x & 4\sin 2x \\ \sin^2 x & \cos^2 x & 1 + 4\sin 2x \\ \hline \textbf{Maxm value of } (2 + 4\sin 2x) = 6 \\ \hline \textbf{4.} & \textbf{R}_2 \rightarrow \textbf{R}_2 - \textbf{R}_1, \textbf{R}_3 \rightarrow \textbf{R}_3 - \textbf{R}_1 \text{ and expand} \\ \hline \textbf{5.} & \text{Take } \textbf{x}, \textbf{y}, \textbf{z} \text{ from } \textbf{C}_1, \textbf{C}_2, \textbf{C}_3 \text{ respectively and multiply } \textbf{R}_1, \textbf{R}_2, \textbf{R}_3 \text{ with } \textbf{x}, \textbf{y}, \textbf{z} \\ \hline \textbf{6.} & \left| \begin{array}{c} a^2 & a & 1 \\ \cos nx & \cos(n+1)x & \cos(n+2)x \\ \sin nx & \sin(n+1)x & \sin(n+2)x \\ \end{bmatrix} = a^2 \sin x - a \sin 2x + \sin x \quad (\text{independent of } n) \\ sin nx & sin(n+1)x & sin(n+2)x \\ \hline \textbf{8.} & \textbf{C}_1 \rightarrow \textbf{C}_1, \textbf{C}_2, \textbf{C}_3 \\ \hline \textbf{8.} & \textbf{C}_1 \rightarrow \textbf{C}_1, \textbf{C}_2, \textbf{C}_3 \\ \hline \textbf{8.} & \textbf{C}_1 \rightarrow \textbf{C}_1, \textbf{C}_2, \textbf{C}_3 \\ \hline \textbf{9.} & \textbf{9} (\textbf{x}) = \left| \begin{array}{c} a^3 + 4x & x + 3 & x - 2 \\ 0 & q - r & r - p \\ \hline \textbf{9.} & \textbf{P}(x) = \left| \begin{array}{c} x^3 + 4x & x + 3 & x - 2 \\ x - 2 & 5x & x - 1 \\ x - 3 & x + 2 & 4x \\ \end{array} \right|$$

$$\begin{split} P(0) &= f = \begin{vmatrix} 0 & 3 & -2 \\ -2 & 0 & -1 \\ -3 & 2 & 0 \end{vmatrix} = -3(-3) - 2(-4) = 17 \\ 10. \quad C_1 \rightarrow C_1 + C_2 + C_3 \\ &\qquad \begin{vmatrix} b^{-}c^{-} & c^{-}a^{-}a^{-}a^{-}b^{+} \\ b^{+}-c^{+} & c^{+}-a^{+}a^{-}b^{+} \end{vmatrix} = 0 = m \\ &\qquad b^{+}-c^{+} & c^{+}-a^{+}a^{-}b^{+} \end{vmatrix} = 0 = m \\ 11. \quad D_1 &= \begin{vmatrix} 1 & 1 & 1 \\ x^2 & y^2 & z^2 \\ x & y & z \end{vmatrix} = \frac{\begin{vmatrix} xyz & xyz \\ xyz \\ 1 & 1 & 1 \end{vmatrix} \\ &\qquad = \begin{vmatrix} yz & xz & xyz \\ xyz & z \end{vmatrix} = 0 \\ 12. \quad \begin{vmatrix} a+b+2c & a & b \\ c & b+c+2a & b \\ c & a & a+c+2b \end{vmatrix} = 2(a+b+c)^{3} [C_1 \rightarrow C_1 + C_2 + C_3] \\ 13. \quad \begin{vmatrix} 0 & 0 & 1 \\ -1 & 1 & 1 \\ -1 & 1 & 2 \end{vmatrix} = 1 = [Z] \\ 14. \quad R_2 \rightarrow R_2 + R_1 \\ R_3 \rightarrow R_3 + R_1 \\ 4(Determinant) = 4 \\ 15. \quad & \Delta_1 = \begin{vmatrix} a_1 & 1 & 1 \\ 0 & b_2 & 1 \\ 0 & 0 & c_3 \end{vmatrix} \begin{vmatrix} a_1 & a_2 & a_3 \\ b_1 & b_2 & b_3 \\ 0 & 0 & c_3 \end{vmatrix} \begin{vmatrix} a_1 & a_2 & a_3 \\ b_1 & b_2 & b_3 \\ 0 & 0 & c_3 \end{vmatrix} \begin{vmatrix} a_1 & a_2 & a_3 \\ b_1 & b_2 & b_3 \\ 0 & 0 & c_3 \end{vmatrix} \begin{vmatrix} a_1 & a_2 & a_3 \\ b_1 & b_2 & b_3 \\ 0 & 0 & c_3 \end{vmatrix} \begin{vmatrix} a_1 & a_2 & a_3 \\ b_1 & b_2 & b_3 \\ 0 & 0 & c_3 \end{vmatrix} \begin{vmatrix} a_1 & a_1 & a_1 \\ a_2 & \beta^2 & \gamma^2 \end{vmatrix}$$

$$\boxed{\begin{array}{l} = \frac{\alpha\beta\gamma}{(1-\alpha)(1-\beta)(1-\gamma)} \left| \begin{matrix} 1 & 1 & 1 \\ \alpha & \beta & \gamma \\ \alpha^2 & \beta^2 & \gamma^2 \end{matrix} \right|} \\ = \frac{\alpha\beta\gamma}{(1-\alpha)(1-\beta)(1-\gamma)} [(\beta-\gamma)(\gamma-\alpha)(\alpha-\beta)] = \frac{-25d}{2(a+b+c+d)} \\ 17. \quad |A|^{3015} |A^2 - 51| \\ 18. \quad R_1 \rightarrow R_1 - R_3 \\ \text{First two rows would be identical} \\ 19. \quad R_3 \rightarrow R_3 - R_2 \\ \Delta = 0 \\ 20. \quad \text{directly evaluate} \\ 21. \quad (1+\cos x)(\sin^2 x) = x^2 + \frac{1}{x^2} \\ \text{This can't be equal anytime as } x^2 + \frac{1}{x^2} \ge 2 \text{ occurs at } x = \\ \text{And } 1 + \cos x = 2, \sin^2 x = 1 \text{ can't be true simultaneously} \\ 22. \quad \sin^{50} x = 1 + \cos^{50} x \\ \sin x = \pm 1; \cos x = 0 \\ X = n\pi \pm \frac{\pi}{2} (n \in N) \\ 23. \quad \left(\frac{\sqrt{3}-1}{2\sqrt{2}}\right) \sin \theta + \left(\frac{\sqrt{3}+1}{2\sqrt{2}}\right) \cos \theta = \frac{1}{\sqrt{2}} \\ \Rightarrow \sin(\theta + 75^\circ) = \sin 45^\circ \\ \Rightarrow \cos(15^\circ - \theta) = \cos 45^\circ \\ \Rightarrow 15^\circ - \theta = 2n\pi \pm \frac{\pi}{4} \\ \Rightarrow \theta = 2n\pi \pm \frac{\pi}{4} + \frac{\pi}{12} \\ 24. \quad \sin^8 x + \cos^8 x = (\sin^4 x + \cos^4 x)^2 - 2\sin^4 x \cos^4 x \\ \Rightarrow \frac{17}{32} = 1 - (\sin 2x)^2 + \frac{(\sin 2x)^4}{8} \end{aligned}$$

 $x = \pm 1$ 

$$\Rightarrow 4(\sin 2x)^4 - 32(\sin 2x)^2 + 15 = 0$$
  

$$\Rightarrow (\sin 2x)^2 = \frac{1}{2}$$
  

$$\Rightarrow 2x = n\pi \pm \frac{\pi}{4}$$
  

$$\Rightarrow x = \frac{n\pi}{2} \pm \frac{\pi}{8}$$
  
25. Use Graph (Total 6 points)  

$$= 2x = \frac{1}{2} + \frac{\pi}{8}$$

26. 
$$\frac{1}{\sin\theta\sin\left(\theta+\frac{\pi}{4}\right)} = \sqrt{2} \left[\cot\theta - \cot\left(\theta+\frac{\pi}{4}\right)\right]$$
$$\sum_{m=1}^{6} \csc\left(\frac{\theta+(m-1)\pi}{4}\right) \csc\left(\frac{\theta+m\pi}{4}\right) = \sqrt{2} \left[\cot\theta - \cot\left(\theta+\frac{3\pi}{2}\right)\right] = 4\sqrt{2}$$
$$\Rightarrow \cot\theta + \tan\theta = 4(2 \text{ solutionsin}\left(0,\frac{\pi}{2}\right))$$

27.  $\theta$  must lie in 4<sup>th</sup> quadrant  $\Rightarrow 2n\pi + \frac{7\pi}{4}$ 

28.  $\cos \theta \cos 2\theta \cos 3\theta = \frac{1}{4}$   $\cos 4\theta = 0 \text{ or } \cos 2\theta = \frac{-1}{2}$  $\theta = \frac{\pi}{8}, \frac{3\pi}{8}, \frac{5\pi}{8}, \frac{7\pi}{8}, \frac{2\pi}{3}, \frac{\pi}{3}$ 

29. 
$$b^2 - 4ac \ge 0$$
  
 $\cos^2 p - 4(\cos p - 1)\sin p \ge 0$   
 $p \in (0, \pi)$ 

30. 
$$1 + \sin \theta \cos \theta - \cos \theta |\sin \theta| - 2 = -1$$
  
 $\sin \theta \cos \theta - \cos \theta |\sin \theta| = 0$ 

$$\Rightarrow \theta \in \left(\frac{3\pi}{2}, \pi\right) \text{ as } \sin \theta - \cos \theta \neq 0 \text{ \& all terms should be finite.}$$

# **PHYSICS**

31. 
$$A = \int_{0}^{R} y \, dx = \frac{2v^{4}}{3g^{2}} \sin^{3} \theta \cos \theta$$
  
32. 
$$v = \sqrt{2g(h-y)}$$
  

$$x = vt = 2\sqrt{y(h-y)}$$
  
33. 
$$\int_{0}^{t} (v-u\cos\alpha) \, dt = l, \int_{0}^{t} v\cos\alpha \, dt = u\tau$$
  
34. 
$$\sin\theta = \frac{1}{\sqrt{2+2gh/v^{2}}}$$
  
35. 
$$\overrightarrow{V_{A}} = 10\hat{i}$$
  

$$\overrightarrow{V_{B}} - \overrightarrow{V_{A}} = 6\cos\theta\hat{i} + 6\sin\theta\hat{j}$$
  

$$\overrightarrow{V_{C}} - \overrightarrow{V_{B}} = -12\hat{i}$$

36. Plot  $(V_A - V_B)$ - t graph Find time till area is positive.

37. 
$$V_{avg} = \frac{displacement}{time}$$

Time is same for all the particles.

Displacement is distance between vertex and incentre.

#### 38.



For minimum relative velocity,  $V \cos \alpha = at$ 

39 Solve the problem by taking OB as X-axis and OA as Y-axis Along OB,  $a = -g \sin \beta$ 

Perpendicular to OB,  $a = -g \cos \beta$ 

41. 
$$\tan 60^{\circ} = \frac{v_C}{v_C}$$

$$\therefore \qquad v_H \\ v_C = v_H \tan 60^\circ = 15m / s$$



42. Let the height of point P above the ground be h.

$$\frac{u\cos 53^{\alpha}t + h = \frac{1}{2}\alpha t^{2}}{\frac{u\sin 53^{\alpha} - zt}{u\cos 53^{\alpha}} = 1}$$

$$h = u\sin 53^{\alpha}t - \frac{1}{2}gt^{2}$$
43.  $T_{i} = \frac{2u\sin(\alpha + \beta)}{g\cos \alpha}$ 

$$T_{i} = \frac{2u\sin(\beta - \alpha)}{g\cos \alpha}$$

$$T_{i} = \frac{2u\sin(\beta - \alpha)}{g\cos \alpha}$$

$$\frac{T_{i}}{T_{i}} = \frac{\sin(\beta + \alpha)}{\sin(\beta - \alpha)}$$

$$\beta - \alpha = \theta$$

$$\sin(\alpha + \beta) - \cos \theta$$

$$\alpha + \beta = 90^{\alpha} - 0$$
 but
$$\beta - \alpha = \theta$$

$$\therefore \beta = 45^{\alpha}$$
44. Relative velocity of balls must be along their line joining  
i.e.  $v\sin 53^{\alpha} = 24\sin 37^{\alpha}$ 
45.  $t = \frac{C}{\mu}, v = \frac{2v_{0}}{c}y$ 
Where  $y$  is the distance from bank from A, and upto middle.  
It follow from this relation that boat moves parallel to banks  
with constant acceleration.  

$$a = \frac{2v_{0}\mu}{C}(v_{0} = at; = a\frac{C}{2\mu}up to middle of the bank)$$

$$\therefore$$
 The boat will reach the middle of the river in time  $T = \frac{C}{2\mu}$ 
and will be carried down stream by  $S = \frac{1}{2}aT^{2} = \frac{V_{0}C}{4\mu}$  when moving from the middle of  
the river to the opposite bank, the boat will again be carried away by a distance of  $\frac{v_{0}C}{4\mu}$ .  
Total dirft  $= \frac{v_{0}C}{2\mu}$ 
46. Conceptual
47. Sides are shown

$$\theta = \pi - 2 \left( \frac{\pi - 2\pi}{n} \right)$$

$$= \frac{2\pi}{n}$$
Here n = 12
$$\Rightarrow \theta - \frac{2\pi}{12} = \frac{\pi}{6}$$
Velocity of approach = V - V cos  $\frac{\pi}{6} = V - \frac{\sqrt{3}}{2} V$ 
Time of approach =  $\frac{u}{V(1 - \frac{\sqrt{3}}{2})}$ 
Distance travelled =  $\left(\frac{2a}{2 - \sqrt{3}}\right)$ .
  
48.  $\bar{v}_{n_{c}} = \bar{v}_{n_{a}} + \bar{v}_{n_{c}} = 4\hat{i} + 2\hat{j}$ 
 $\bar{v}_{n_{a}} = \hat{o}i$ 
 $\bar{v}_{n_{a}} = \hat{v}_{n_{a}} - \hat{v}_{n_{a}} = 6\hat{i} - 4\hat{i} - 2\hat{j} = 2\hat{i} - 2\hat{j}$ 
Direction will be north-west
  
49.  $Vp/w = 500 \text{ kmph} \hat{j}$ 
 $Vy/g = v_{s}\hat{i} + v_{s}\hat{j}$ 
 $Vp/g = v_{s}\hat{i} + v_{s}\hat{j}$ 
 $Vp/g = v_{s}\hat{i} + v_{s}\hat{j}\hat{j} = 150\sqrt{2}\hat{i} + 1150\sqrt{2}\hat{j} \text{ km} = \frac{2v_{s}\hat{i} + (2v_{s} + 1000)\hat{j}}{2}$ 
 $\Rightarrow vx = 75\sqrt{2}$ 
 $Vw/g = 75\sqrt{2}\hat{i} + 75\sqrt{2}\hat{j} = 150 \text{ kmph at 45^{\circ} N \text{ of E}}$ 
  
50. Net velocity of boat in river =  $\sqrt{5^{2} - u^{2}}$ 
 $t = \frac{Distance}{Velocity} \Rightarrow \frac{1}{4} = \frac{1}{\sqrt{5^{2} - u^{2}}} \Rightarrow u = 3kmh^{-1}$ 
  
51. For the minimum time man should swim perpendicular to the direction of flow of stream.
  
52. Velocity of boat w.r.t. ground is  $\vec{v} = (u - vain0)\hat{i} + vcas\hat{0}$ 
For the required condition  $v_{0} = u - vsin\theta$ 
  
53.  $y = x tan \theta - \frac{gx^{2}}{2u^{2} \cos^{2} \theta}$ 
Let  $m = tan \theta$ 
 $\left(\frac{gx^{2}}{2u^{2}}\right)m^{2} - xm + \left(\frac{gx^{2}}{2u^{2}} + y\right) = 0$ 

Let 
$$\overrightarrow{V_r} = a\hat{i} + b\hat{j}$$
  
 $\overrightarrow{V_M} = 3\hat{i} + \sqrt{3}\hat{j}$   
 $\overrightarrow{V_R} - \overrightarrow{V_m}$  is along vertical  
 $\therefore a = 3$   
 $\sqrt{a^2 + b^2} = 5$   
 $b = 4$   
Conceptual

59. Conceptual60. Conceptual

### CHEMISTRY

1	
61.	$\Delta Hr = 2(\Delta H_f)SO_2 + (\Delta H_f)CO_2 - (\Delta H_f)CS_2$
	$-275 = 2 \times (-75) + x - 25, x = -100$
62.	It $\Delta ng = 0$ then $K_P = K_C$
	On increasing pressure reaction will proceed in the direction in which lesser number of
	gaseous moles are formed
	In homogeneous equilibrium all the species are in single phase
63.	Conceptual
64.	$6PCl_5 \Rightarrow PCl_3 + Cl_2$
	teq $1 - \alpha$ $\alpha$ $\alpha$
	$6\mathrm{Kp} = \frac{\alpha^2}{1-\alpha^2} \times P$
	$6 = \frac{\frac{1}{16}}{1 - \frac{1}{16}} \times 30$
65.	$K_p = P_{CO2}$
	In hetrogeneous system equilibrium is unaffected by small amount addition of solid or
	liquid
66.	Conceptual
67.	Conceptual
68.	Conceptual
69.	$\Delta H = \Delta E + \Delta n_g RT = 2.1 + 2 \times 0.002 \times 300 = 3.03 kcal$
	$\Delta G = DH - T\Delta S = 3.3 - 300 \times (0.02) = -2.7kcal$
70.	Conceptual
71.	Conceptual
72.	Conceptual
73.	Conceptual
74.	Conceptual
75.	as $\Delta n_g < O$ and $\Delta H < O$ hence reaction is favourable at high pressure and low
	temperature.
76.	As density of ice is less than density of water hence on increasing pressure some more
	ice will melt.
	Page 10

77. Conceptual  
78. 
$$\frac{2SO_{3}(g) \rightleftharpoons 2SO_{2}(g) + O_{2}(g)}{1-\alpha} = \frac{a}{a} = \frac{a}{2}$$

$$K_{p} = \frac{\left(\frac{a/2}{1+\alpha/2}, P_{p}\right)^{2}}{\left(\frac{1-\alpha}{1+\alpha/2}, P_{p}\right)^{2}} = \frac{a}{(2+\alpha)} \times \frac{P_{e} \times a^{2}}{(1-\alpha)^{2}}$$

$$CH_{3}COOH(\ell) + C_{2}H_{3}OH(\ell) \rightleftharpoons CH_{3}COOC_{2}H_{3}(\ell) + H_{2}O(\ell)$$
79. 
$$3 \quad 3 \quad 0 \quad 0$$

$$3-2 \quad 3-2 \quad 2 \quad 2$$

$$K_{c} = \frac{2\times2}{1\times1} = 4$$
80. Conceptual  
81. Conceptual  
82. Conceptual  
83. 
$$K_{p} = (P_{u_{2}0})^{2}$$

$$P_{u_{3}0} = \sqrt{K_{p}}$$

$$= \sqrt{2.25 \times 10^{-6}}$$

$$= 15 \times 10^{-3}$$

$$R. H = \frac{P_{u_{2}0}}{SV.P} = \frac{15 \times 10^{-3}}{22.8/760}$$

$$= \frac{15 \times 10^{-3}}{3 \times 10^{-2}} = 0.5$$

$$R.H = 0.5 \times 100 = 50\%$$
84. Conceptual  
85. 
$$(P_{u_{2}0})_{1} = (5 \times 10^{-12})^{\frac{1}{2}} = (5)^{1/4} \times 10^{-3}$$

$$(P_{u_{2}0})_{2} = (2.43 \times 10^{-13})^{\frac{1}{5}} = 3 \times 10^{-3}$$

$$(P_{n_2o})_3 = (1.204 \times 10^{-37})^{\frac{1}{0}} = 2 \times 10^{-3}$$
  
as  $(P_{n_2o})_1$  is least hence SrCl<sub>2</sub>.2H<sub>2</sub>O will act as good dehydrating agent  
 $2NH_3(g) \rightleftharpoons N_3(g) + 3H_2(g) K_p = 27 \times 10^{-8} P_0^2$   
86.  
 $1 - \alpha$   
 $\frac{\alpha}{2}$   
 $\frac{3\alpha}{2}$   
 $n_7 = 1 + \alpha$   
 $27 \times 10^{-8} P_0^2 = \frac{\left[\frac{\alpha}{2(1+\alpha)}P_0\right] \left[\frac{3\alpha}{2(1+\alpha)}P_0\right]^3}{\left[\frac{1-\alpha}{1+\alpha}P_0\right]^2}$   
 $27 \times 10^{-8} = \frac{3^3 \alpha^4 P_0^5 P_0^2}{2 \times 2^3 \times P_0^2}$   
 $2^4 \times 10^{-8} = \alpha^4$   
 $\alpha = 2 \times 10^{-2}$   
87.  
 $\Delta G^{\circ} = -RT \ln k$   
 $= -2 \times 300 \times \ell n e^4$   
 $= -2 \times 300 \times 4 \times 10^{-3}$   
 $= -2.4k cal.$   
88.  
Conceptual  
89.  
Conceptual  
90.  
PCl<sub>3</sub>  $\rightleftharpoons PCl_3 + Cl_3$   
 $1-\alpha$   
 $\alpha$   
 $M_{mx} = \frac{M}{1+\alpha}$   
 $d = \frac{PM_{mx}}{RT}$   
 $\Rightarrow d = \frac{PM}{RT}$   
 $\Rightarrow \alpha = \frac{PM}{dRT} - 1$